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# Diode Laser-Based Measurements of Hydrogen Fluoride Gas During Chemical Suppression of Fires

by Kevin L. McNesby, R. Reed Skaggs,  
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## **Diode Laser-Based Measurements of Hydrogen Fluoride Gas During Chemical Suppression of Fires**

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## Abstract

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Near-infrared tunable diode laser (NIR-TDL) spectroscopy is used to quantify hydrogen fluoride (HF) gas produced during fire-suppressant testing of Halon alternatives. Results of comparisons with other techniques for measuring HF gas concentrations are discussed. Measurements of HF gas produced in laboratory- and real-scale fire-suppression testing are presented. The necessity for time-resolved measurements during testing of suppression systems designed to scavenge HF gas is demonstrated.

## **Acknowledgments**

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# 1. Introduction

Fire and explosion suppression is of critical importance onboard land combat vehicles. Currently, Halon 1301 ( $\text{CF}_3\text{Br}$ ) is employed to suppress fires and explosions in most military vehicles. However, due to its high ozone depletion potential (ODP), Halon 1301 was banned from production by international agreement [1] as of 1 January 1994. Currently, a service-wide program\* is underway to identify near-term environmentally acceptable Halon alternative technologies.

Important properties for successful Halon alternatives are high-suppression efficiency, low ODP, low-suppressant residue after use, low electrical conductivity, stability under long-term storage, and nontoxicity prior to and during dispersion. Recent work [2] has shown that hydrogen fluoride (HF) gas is the principal toxic gas produced during fire suppression by Halon 1301 and by fluorocarbon-based fire suppressants. Therefore, for fluorocarbon-based Halon alternatives, minimization of HF gas production during fire suppression is an important criterion. The work presented here describes our efforts using near-infrared tunable diode lasers (NIR-TDL) to measure HF gas produced during fluorocarbon-based chemical suppression of laboratory- and real-scale fires.

## 2. Background

When a projectile penetrates the hull and fuel cell of a land combat vehicle, fuel-cell penetration may result in a fine mist of fuel being dispersed into the crew compartment. The ensuing "mist fireball explosion" [3] is extinguished within 250 ms of detection, using the currently deployed Halon 1301 suppression system. Diagnostic equipment designed to measure HF gas produced during testing of Halon alternative systems needs to provide data on a time scale shorter than the suppression event. The extreme reactivity of HF gas dictates the additional requirement that the measurement be made in situ.

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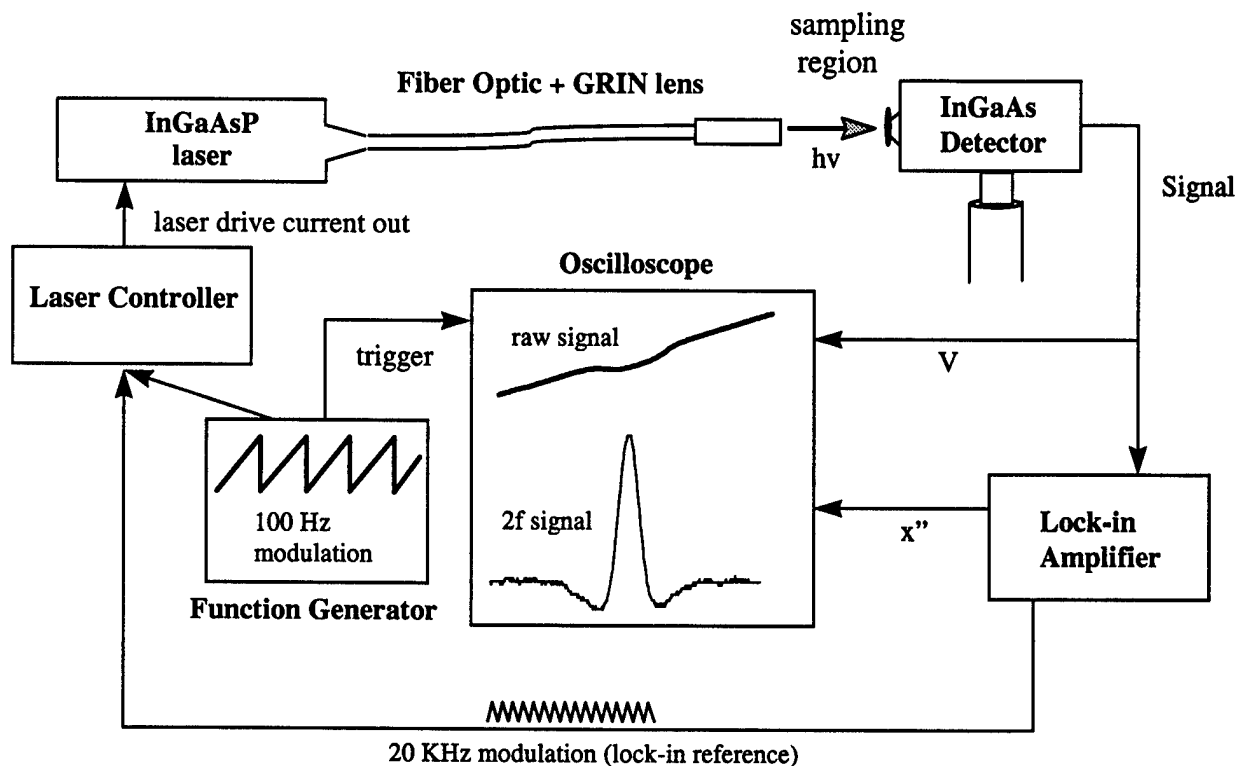
\* Next-Generation Fire-Suppression Technology Program, administered by the Department of Defense (DOD).

For HF measurement, NIR-TDL absorption spectroscopy provides the midrange sensitivity (50–5,000 ppm), speed of data collection, and ability to use low replacement-cost optics and detectors necessary for data sampling in hostile environments. Relatively low-cost narrow band diode lasers emitting NIR radiation at wavelengths absorbed by HF gas ( $\sim 1.3\text{-}\mu\text{m}$  wavelength-overtone transition) are commercially available. Diode lasers emitting in the mid-infrared region at wavelengths where HF absorbs ( $2.5\text{-}\mu\text{m}$  wavelength-fundamental transition) are not yet commercially available.

### 3. Experimental

Figure 1 shows a schematic of the experimental setup and signal-processing electronics. The experimental apparatus does not appreciably change according to the type of fire being investigated. For experiments described here, an InGaAsP distributed feedback (DFB) NIR-TDL (Sensors Unlimited, Inc.) operating near  $7,665\text{ cm}^{-1}$  (corresponding to P(2)) of the first vibrational overtone of HF [4]) is mounted in a temperature-controlled laser diode mount (ILX Lightwave model LDM-4980). The temperature of the diode mount is controlled by a thermoelectric temperature controller (ILX Lightwave model LDT-5412). Laser diode current and internal thermoelectric cooler (TEC) control are provided using a laser diode controller (ILX Lightwave model 3700). A low-frequency sawtooth modulation ( $\sim 100\text{ Hz}$ , Tektronix FG 504 Function Generator) is combined with a higher frequency modulation ( $\sim 20\text{ KHz}$ , sine output from an SRS Model 830 DSP lock-in amplifier). This modulation is superimposed onto the laser drive current, allowing the laser wavelength to be scanned through the spectral region of interest (low-frequency modulation) while also providing a high frequency wavelength “dither” to allow second harmonic signal detection.

The wavelength modulated radiation from the laser diode is fiber-coupled. This fiber is terminated by a gradient index (GRIN) lens (Sentech Systems, Inc.), which collimates the laser radiation. The GRIN lens-tipped fiber is placed into a small test rig designed to maintain fiber and detector alignment. Laser radiation is detected using an InGaAs detector. The path length from grin lens to detector is typically 10 cm. The modulated detector signal is demodulated at twice the higher



**Figure 1. A Schematic of the Experimental Apparatus and Signal-Processing Electronics Used in These Experiments.  $V$  Describes the Signal Prior to Demodulation, and  $x''$  Describes the Signal Following Demodulation at Twice the Reference Frequency.**

frequency modulation (SRS Model 830 DSP lock-in amplifier), the second derivative line shape is displayed on an oscilloscope (LeCroy Model 9354), and the output from the oscilloscope is stored on a laptop computer.

## 4. Measurements in Particulate-Matter-Laden Environments

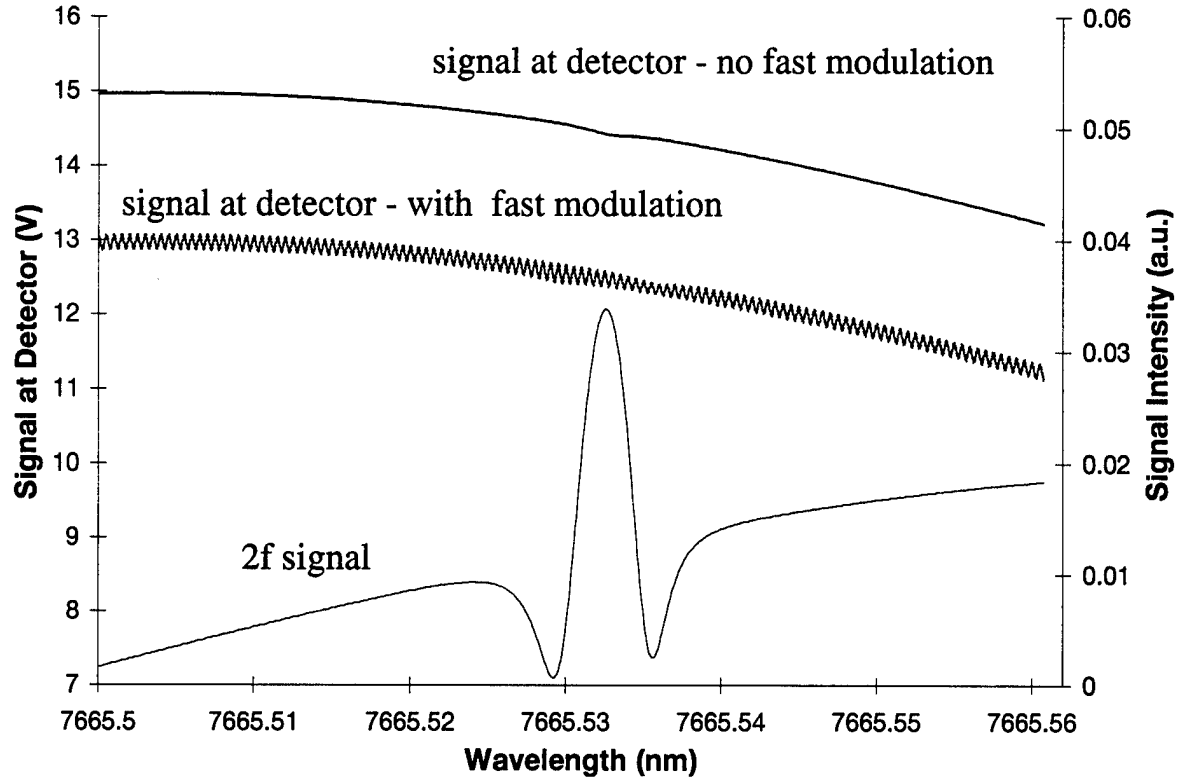
Normal fixed wavelength transmission spectroscopy cannot distinguish between light attenuation (measured at the detector) resulting from gas absorption or light scattering such as might occur in a smoke- or soot-filled environment. Partly for this reason, we employ derivative spectroscopy to exclude contributions to light attenuation by scattering from particulate matter. In this method, the laser output wavelength is scanned through a wavelength range, which includes the wavelength of

absorption of the gas species of interest (HF). A lock-in amplifier is used to measure the change in detector signal intensity with the change in wavelength. Attenuation of laser radiation by a ro-vibrational transition of a small gas molecule is detected because the wavelength range of the scan is several times the width of the spectral absorption feature (typically on the order of  $0.1 \text{ cm}^{-1}$  at atmospheric pressure). Because light scattering by particulate matter is nearly constant over the very small wavelength range of the laser scan, the change in detector signal intensity with the change in wavelength is effectively zero in the absence of any absorbing gas.

Second harmonic detection using TDLs has been extensively discussed in the literature [5]. Figure 2 shows a graphical simulation of the signal processing employed in these experiments. Briefly, the laser output wavelength is slowly scanned through a spectral region where HF gas absorbs (in this case, P(2) of the first vibrational overtone). The absorption in the absence of any modulation may be seen in the upper trace of Figure 2. A small-amplitude high-frequency modulation is superimposed on the laser drive current (middle trace in Figure 2). Demodulation at twice the frequency of the small-amplitude high-frequency laser drive modulation yields the second derivative signal shown in the lower trace in Figure 2. It should be noted that the upper trace in Figure 2 shows that the laser diode output power is a nonlinear function of wavelength. While this is nonideal behavior, in our experience, this is a common trait of commercially available laser diodes. The nonlinear power dependence on wavelength causes the sloping baseline (exaggerated here for illustration) for the second derivative signal in Figure 2. For measurements at extremely low concentrations or for gases with small absorption cross sections, laser diode output power dependence on wavelength may affect limits of detection.

## 5. Calibration

Calculation of HF gas concentration relies upon the Bouguer-Lambert Law [6], which states that for optically thin media at the absorption line center, the negative logarithm of the fraction of transmitted light intensity (referred to as absorbance,  $A$ ) is equal to the absorption coefficient of the molecule of interest ( $\alpha$ ) times path length ( $L$ ) times pressure ( $P$ ):  $A = \alpha LP$ . Since the absorption



**Figure 2. A Graphical Simulation of Signal Processing for 2f Measurement of HF Gas Concentrations. The Slope in the Baseline Is Caused by the Change in Laser Diode Output Power With Wavelength.**

coefficient ( $\alpha$ ) is constant and  $L$  is known, a measurement of absorbance ( $A$ ) is sufficient to determine pressure ( $P$ ). The second derivative signal peak height may be shown [7] to be proportional to absorbance ( $A$ ):

$$x''/V = kA. \quad (1)$$

Here,  $x''$  is the peak height (peak to trough) of the second derivative signal (volts),  $V$  is the direct-current (DC) voltage measured by the detector in the absence of any absorbance, and  $k$  is a constant that includes the measuring instrument and optics function. Letting  $S$ , denoted as 2f signal peak height (peak to trough), equal  $x''/V$  yields

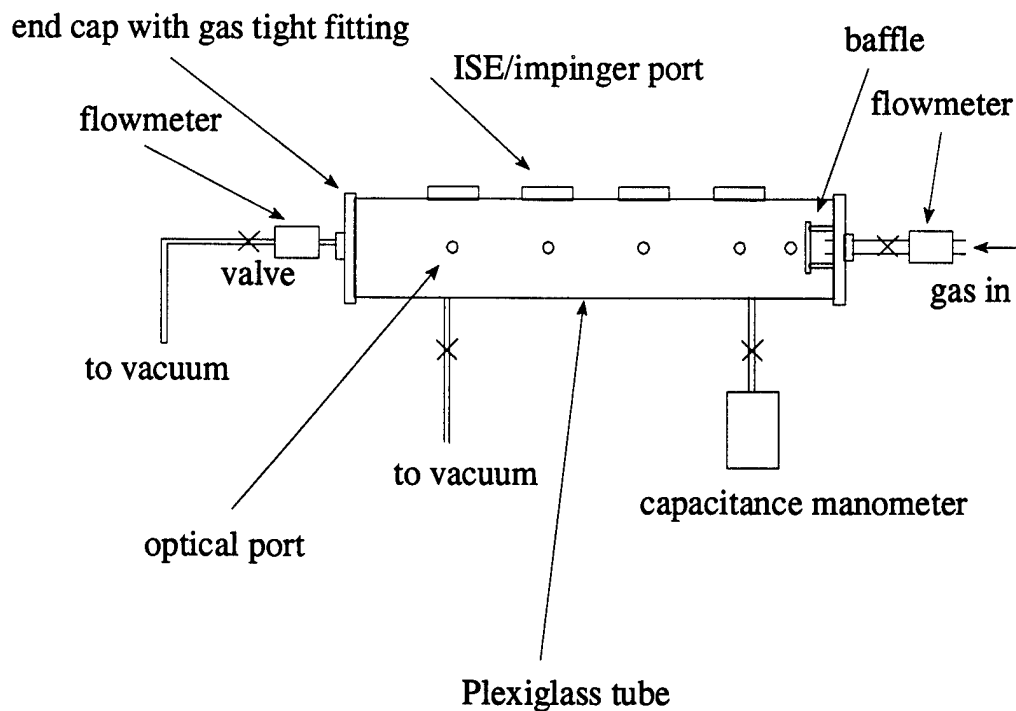
$$S = (k\alpha) LP. \quad (2)$$

The slope of a plot of LP vs. S provides the value of  $k\alpha$ . A calibration of the system using known concentrations of the absorbing gas must be performed to determine the value of  $k\alpha$ . Once this value is known, the HF gas pressure (P) may be obtained directly from equation (2). Care must be exercised so that HF degradation of optical surfaces during measurement does not affect the value of  $k\alpha$ , since this value is instrument-function-dependent. For this reason, calibration should be performed at the beginning and end of each measurement set. Additionally, for HF concentrations that attenuate more than 5% of the incident light, the linear relationship between HF gas pressure and 2f signal peak height may no longer hold [7].

## 6. Comparison With Other Techniques

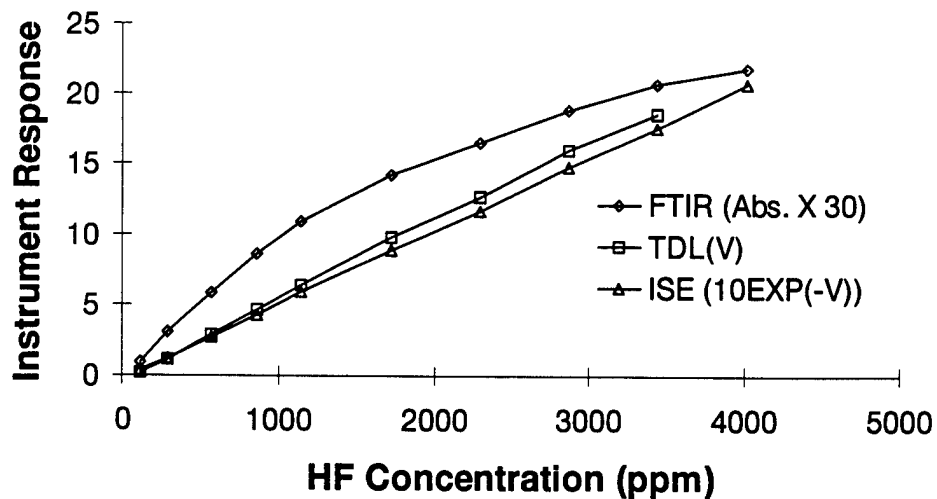
A flow cell was constructed to allow simultaneous measurement of HF gas mixtures (HF in N<sub>2</sub>) using NIR-TDL absorption spectroscopy, Fourier transform infrared (FTIR) spectroscopy, and using a fluoride ion selective electrode (ISE). The flow cell is designed with five radial sampling ports at four longitudinal distances from the gas inlet port. All measurements were made at atmospheric pressure. A schematic of the flow cell is shown in Figure 3. Results from measurements of HF gas using the three techniques are shown in Figure 4. The limit of detection (LOD) of HF gas using the NIR-TDL absorption spectroscopy apparatus described here is approximately 40 ppm. Calibration of the FTIR data used the peak absorbance of the R(3) line of the fundamental ( $1 \leftarrow 0$ ) vibrational transition. The limit of detection of HF gas using FTIR absorption spectroscopy was approximately 10 ppm. The absorption techniques give similar limits of detection because the increased signal-to-noise ratio achieved using phase-sensitive detection with the NIR-TDL ( $2 \leftarrow 0$ ) is offset by the higher absorption coefficient for the fundamental transition measured using FTIR absorption spectroscopy. The nonlinearity of the FTIR absorbance vs. concentration data is caused by an interpolation error that occurs because the instrument resolution ( $0.5 \text{ cm}^{-1}$ ) is large compared to the linewidth being measured. Fitting the entire rovibrational envelope of the FTIR spectrum results in improved linearity. The limit of detection for HF gas using ISE was approximately 1 ppm. It should be pointed out that the ISE measures total fluoride ion concentration, so the technique is unable to distinguish between HF and CF<sub>2</sub>O, which is produced in appreciable quantities during chemical suppression of large-scale fires [2].





**Figure 3. The Flow Tube Designed to Allow Simultaneous Measurement of HF Gas Using NIR-TDL Absorption Spectroscopy, FTIR Spectroscopy, and Fluoride ISE. There Are Five Feedthrough Ports (One for the ISE, Plus Four Optical Ports) per Longitudinal Position.**

### Instrument Calibration



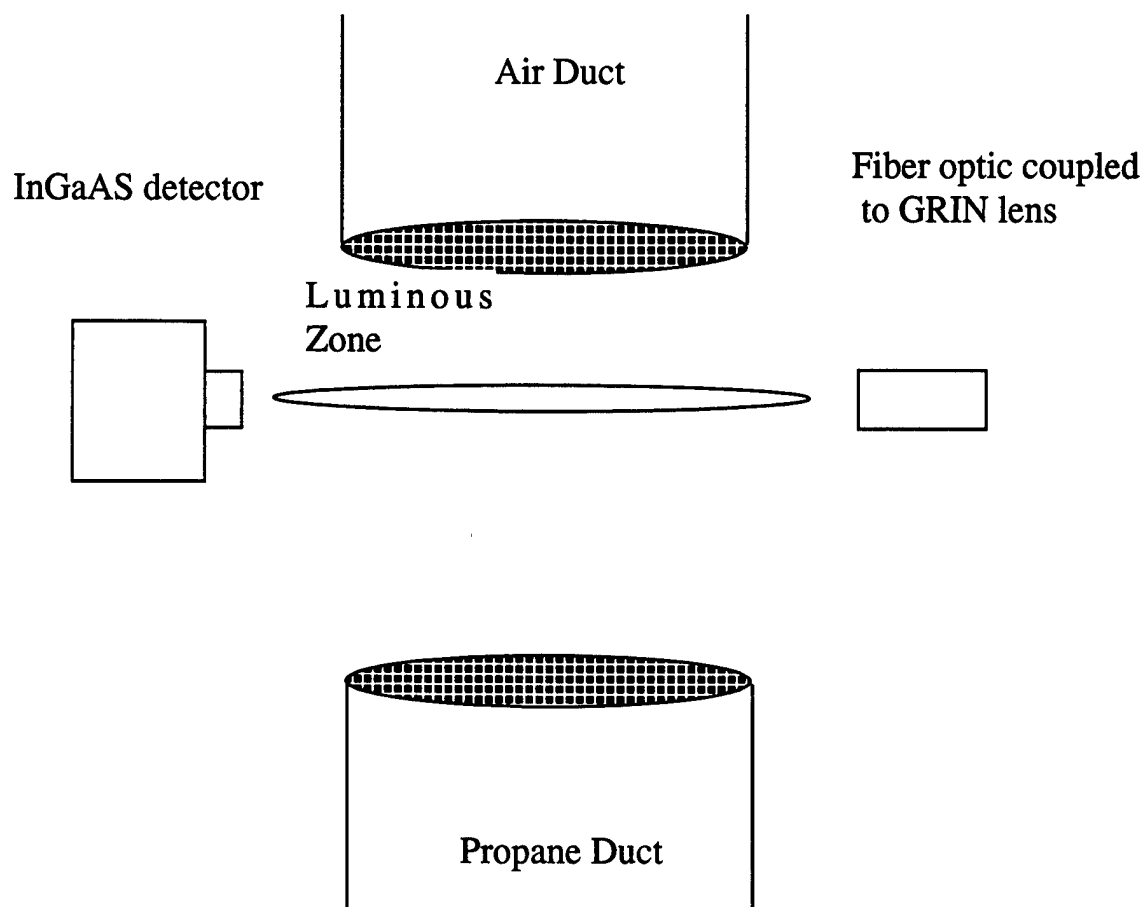
**Figure 4. Results of Comparative Calibration of FTIR, NIR-TDL, and Fluoride ISE for HF Gas Detection.**

## 7. Results and Discussion

Measurements of HF gas concentrations in well-controlled optically thin environments, such as described in the previous section, are necessary to check on instrument performance and calibration but may not reveal difficulties encountered in many laboratory- and real-scale tests. In this section, we describe measurements of HF gas using NIR-TDL absorption spectroscopy in an atmospheric pressure opposed-flow propane/air flame suppressed by  $C_3F_7H$  (trade name FM-200) and in a 30-kW heptane/air pan fire during extinguishment by a hybrid suppression system ( $C_3F_6H_2$  [trade name FE-36] plus particulate) designed to scavenge HF gas.

**7.1 Laboratory Flame-Suppression Studies.** Figure 5 shows a schematic of an opposed-flow burner apparatus and the positioning of the fiber-optic GRIN lens assembly and detector used to measure HF gas produced in a propane/air/fluorocarbon flame. By varying the mass flow rate of fuel and oxidizer for a given flame in the opposed flow burner, the mole fraction of a given suppressant needed for extinguishment may be varied. The ability to vary the “strength” of a flame in the opposed flow burner (strength being inversely proportional to the ratio of diffusion to chemical times, referred to as the Damkohler number) [8] allows flames of different strengths to be investigated, and allows for a ranking of Halon alternatives according to their flame-suppression efficiency.

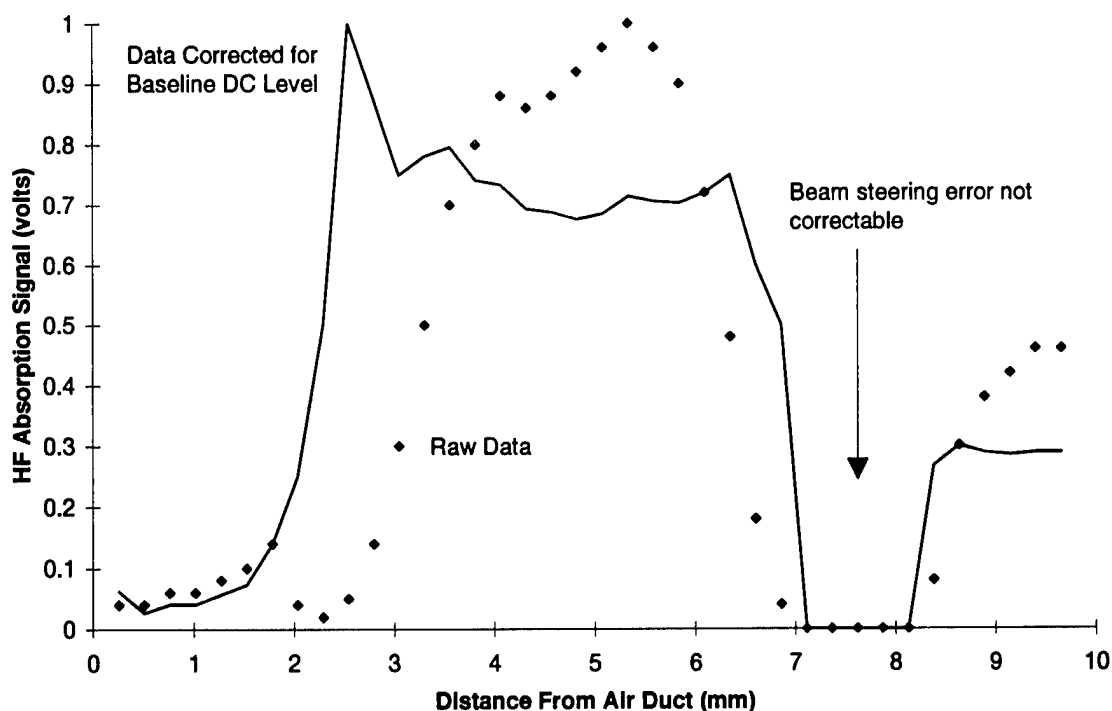
Figure 6 shows the results of NIR-TDL measurement of HF gas produced in a propane/air opposed-flow flame (strain rate =  $50\text{ s}^{-1}$ ) to which 1%  $C_3F_7H$  (trade name FM-200) has been added to the air stream. To generate this data, the fiber-optic GRIN lens assembly and detector were kept fixed in position and the burner assembly was translated vertically. In this figure, raw data (peak to trough distance of the  $2f$  waveform) are represented by symbols, and data corrected for changes in baseline DC level are shown by a line. The change in DC baseline level as different heights in the flame are probed is caused by density gradients that cause beam steering when the line of sight is in the vicinity of the luminous zone of the flame. Using  $2f$  detection, these beam steering-induced errors may be compensated for as long as some laser radiation reaches the detector. The case where



**Figure 5. Schematic of the Atmospheric Pressure Opposed-Flow Burner, Fiber Optic Coupled to GRIN Lens for Delivery of Laser Radiation and InGaAs Detector Element.**

no correction is possible may be seen in the region from 7 to 8 mm from the oxidizer duct. In this region, no laser radiation was incident on the detector.

**7.2 Field Flame-Suppression Studies.** Figure 7 shows results of NIR-TDL measurement of HF gas in a test designed to measure the effect of particulates designed to scavenge HF. For these tests, a fiber-optic GRIN lens and detector assembly (path length = 10 cm) was placed in a 1.5-m<sup>3</sup> cubic enclosure near an air-fed liquid heptane pan fire (fire size approximately 30 kW). After allowing the fire to burn uninhibited for 15 s, the enclosure was flooded with 10% by volume C<sub>3</sub>F<sub>6</sub>H<sub>2</sub> (trade name FE-36), with and without the particulate matter additive. A new data point was

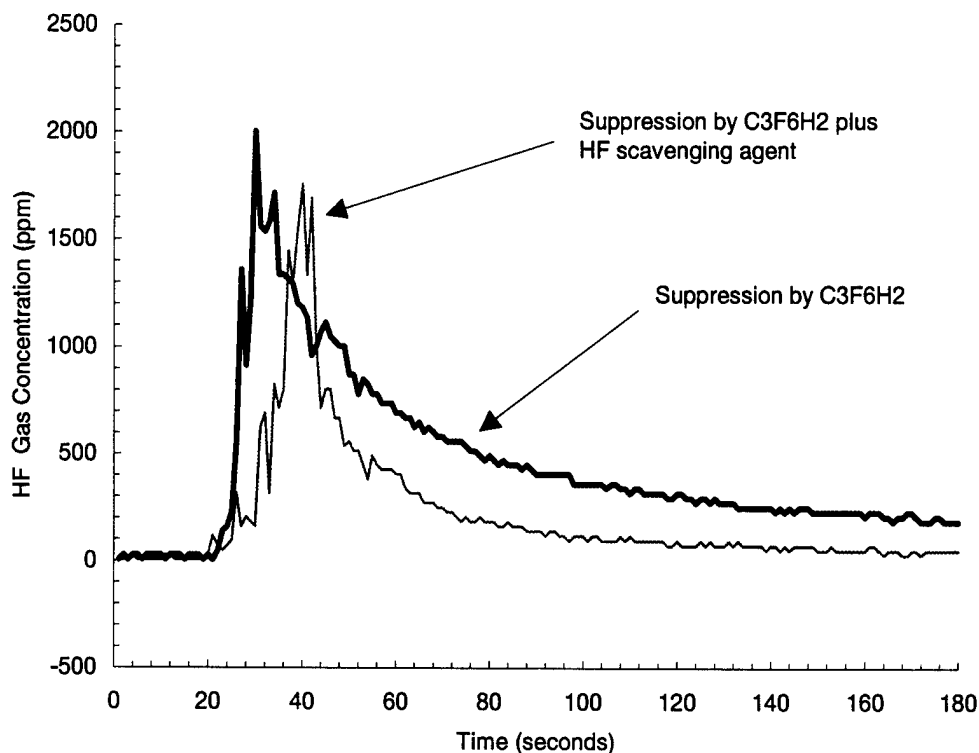


**Figure 6. HF Gas Absorption vs. Distance From the Oxidizer Duct for an Atmospheric Pressure Opposed-Flow Propane/Air Flame to Which 1%  $C_3F_7H$  Has Been Added to the Air Stream. The Symbols Are Raw Data, and the Line Represents the Data Corrected for Fluctuations in DC Baseline Signal Level.**

measured every 10 ms. Figure 7 shows that the rate at which HF gas concentration decreases after peak concentration is reached is fastest for the fire extinguished using FE-36 to which particles have been added. For this test, the measurement indicates that the particulate matter added to the fluorocarbon-based suppressant (FE-36) successfully reduced the time-weighted average HF concentration.

## 8. Conclusion

NIR-TDL laser absorption spectroscopy is an extremely useful tool for measuring HF gas concentrations produced during fire suppression by fluorocarbon-based suppressants. The low cost of fiber optics and detectors allows for real-time in-situ measurement of HF gas in hostile environments. Care must be taken to ensure that calibration is maintained during testing because most optic surfaces are damaged during measurement.



**Figure 7. HF Gas Concentration vs. Time for Heptane Pan Fires Suppressed by  $C_3F_6H_2$  With and Without Addition of HF Scavenging Agent. Note the Difference in the Rate at Which HF Gas Concentration Decreases With and Without Scavenging Agent.**

Future experiments will involve multiplexing of several lasers to enable multiple species to be measured using one fiber and one detector. Additionally, recent advances in laser fabrication have made diode lasers at wavelengths near  $2\ \mu\text{m}$  commercially available. This advance improves limits of detection for many species by decreasing the quantum number change in absorption, as well as enabling measurement of species whose highest lying fundamental has, up to now, been out of reach of commercially available NIR laser diodes.

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## 9. References

1. "The Montreal Protocol on Substances That Deplete the Ozone Layer." 1994 Copenhagen Amendments, 1987.
2. McNesby, K. L., R. G. Daniel, A. W. Miziolek, and S. H. Modiano. "Optical Measurement of Toxic Gases Produced During Firefighting Using Halons." *Applied Spectroscopy*, vol. 51, pp. 678–683, 1997.
3. Finnerty, A. E., and S. Polyanski. *Journal of Fire Science*. Vol. 2, p. 242, 1993.
4. Rothman, L. S., R. R. Gamache, R. H. Tipping, C. P. Rinsland, M. A. H. Smith, D. C. Brenner, V. Malathy Devi, J.-M. Flaud, C. Camy-Peyert, A. Perrin, A. Goldman, S. T. Massie, L. R. Brown, and R. A. Toth. "The Hitran Molecular Database: Editions of 1991 and 1992." *Journal Quant. Spectroscopic Radiat. Transfer*, vol. 48, pp. 469–507, 1992.
5. Reid, J., and D. Labrie. "Second-Harmonic Detection With Tunable Diode Lasers - Comparison of Experiment and Theory." *Applied Physics*, vol. B 26, pp. 203–210, 1981.
6. Guelachvili, G., and K. Rao. *Handbook of Infrared Standards*. Academic Press, Orlando, FL, 1986.
7. Miller, J. H., S. Elreedy, B. Ahvazi, F. Woldu, and P. Hassanzadeh. "Tunable Diode-Laser Measurement of Carbon Monoxide Concentration and Temperature in a Laminar Methane-Air Diffusion Flame." *Applied Optics*, vol. 32, pp. 6082–6089, 1993.
8. Chelliah, H. K., C. K. Law, T. Ueda, M. D. Smooke, and F. A. Williams. "An Experimental and Theoretical Investigation of the Dilution, Pressure, and Flow-Field Effects on the Extinction Condition of Methane-Air-Nitrogen Diffusion Flames." *Twenty-Third Symposium (International) on Combustion*, pp. 503–511, The Combustion Institute, 1990.

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Street or P.O. Box No.

\_\_\_\_\_  
City, State, Zip Code

7. If indicating a Change of Address or Address Correction, please provide the Current or Correct address above and the Old or Incorrect address below.

• OLD  
• ADDRESS

\_\_\_\_\_  
Organization

\_\_\_\_\_  
Name

\_\_\_\_\_  
Street or P.O. Box No.

\_\_\_\_\_  
City, State, Zip Code

(Remove this sheet, fold as indicated, tape closed, and mail.)  
(DO NOT STAPLE)